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Tensile Strength Properties of Rice Husk-Rice Husk Ash Filled Plastic Drinking Bottle Waste Hybrid Composite with the Addition of Glycerol as Plasticizer

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Abstract: This study aims to determine the tensile strength properties of rice husk-rice husk ash filled plastic drinking bottle waste hybrid composite with the addition of glycerol as plasticizer. Hybrid composite was made using extruder at 265 °C by mixing plastic drinking bottle waste with 100 mesh particle size of rice husk and rice husk ash with the ratio between matrix and filler 95/5, 90/10 and 85 / 15 (w/w) and glycerol as a plasticizer added as much as 3% of the volume plastic bottle waste on each ratio, and then it was processed using hotpress at 265 °C for 5 minutes. Results of testing the tensile strength properties indicate that plastic drinking bottle waste hybrid composite filled by rice husk -rice husk ash with the addition of glycerol in the ratio of 95/5 obtained maximum tensile strength of 2.717 MPa, a Young's modulus value of 98 964 MPa, elongation at break of 2.603%. Results of SEM analysis on plastic drinking bottle waste hybrid composite filled by rice husk ash-rice husk with addition of glycerol in the ratio of 95/5 shows spread evenly on the surface of the filler matrix.

Keywords: hybrid composite, rice husk ash, rice husk, glycerol, tensile strength

1. Introduction

Nowadays, material technology has been developed rapidly. It is due to the increase of needs on materials that can meet particular requirement. One of the material technology product is polymer composite material. The easy forming ability, good strength and the advantage on strength : weight ratio, motivate the use of polymer composite as replacement for conventional metal material at various products [1].

Hybrid composite often related with fiber reinforcing material which was based on resin raw material. It was made by joining two kinds of fiber into single matrix. This concept was the simple extention from composite principle which combine two or more material to optimize the selling price by utilizing its best quality while reducing the unwanted effects. The combination of various material can also be called as hybrid [2].

Waste was the consequence of human activity. As the population and life style increased, the waste volume produced was also increased. Most of the waste produced were household waste which contained organic and anorganic materials. From the two categories of waste, it was known anorganic waste had more difficult to handle than the organic waste. The anorganic waste could not be degraded easily by nature and could become a serious land pollution problem. One of the examples of anorganic waste was plastic waste [3].

Plastic waste has become main problem in society and can be found in various places, especially in waste disposal. Plastic waste can cause environmental pollution because of its low biodegradability [4].

Paddy was one of main agricultural products in agricultural nations, especially in Indonesia. Rice husk was abundant byproduct from paddy grinding process. It was used as fuel for combusting red stone, cooking or disposing as such. The inappropriate rice husk handling would cause environmental pollution. Previous research had reported that approximately 20% of paddy weight were rice husk and 13 – 29% of the husk composition were rice husk ash which were produced by combusting the husk. Generally, silica (SiO2) composition value in rice husk ash was 94 – 96%. If this value near or below 90%, it was probably cause by contamination of other substance with low silica composition. Rice husk ash produced from controlled combustion at high temperature (500 – 600 °C) will produce silica ash that can be utilize for various chemical process [5].

In this study, glycerol was chosen as plasticizer because it was one of plasticizers that had been widely used and proved to reduce internal hydrogen bond effectively so that the intermolecular distance would increase [6].

The objective of this study was to determine the composition effect of plastic drinking bottle waste matrix and rice husk-rice husk ash particle with glycerol addition as plastisizer on tensile strength of hybrid composite obtained.

2. Theory

Plastic is polymer material that could not easily decompose by decomposer microorganism. Therefore, used plastic piling will cause environmental problem [9]. Nowadays, plastic waste has become main problem in society and can be found in various places, especially in waste disposal. Plastic waste can cause environmental pollution because of its low biodegradability [4].
The purpose of hybridization is to form new material that maintain the advantages of its constituent and minimize its disadvantages. Hybridization can provide benefits in cost and improvement of the mechanical properties. Therefore, the production cost in producing eco-friendly product can be reduced [7].

Husk is categorized as usable biomass for various needs such as industrial raw material, animal feed, and fuel energy resources. According to the chemical composition data, husk contains various essential chemical elements as shown in Table 1.

<table>
<thead>
<tr>
<th>Component</th>
<th>Weight (%)</th>
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<tbody>
<tr>
<td>Water content</td>
<td>32.40 – 11.35</td>
</tr>
<tr>
<td>Protein</td>
<td>1.70 – 7.26</td>
</tr>
<tr>
<td>Fat</td>
<td>0.38 – 2.98</td>
</tr>
<tr>
<td>Free nitrogen extract</td>
<td>24.70 – 38.79</td>
</tr>
<tr>
<td>Fiber</td>
<td>31.37 – 49.92</td>
</tr>
<tr>
<td>Ash</td>
<td>13.16 – 29.04</td>
</tr>
<tr>
<td>Pentose</td>
<td>16.94 – 21.95</td>
</tr>
<tr>
<td>Cellulose</td>
<td>34.34 – 43.80</td>
</tr>
<tr>
<td>Lignin</td>
<td>21.40 – 46.97</td>
</tr>
</tbody>
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Rice husk ash produced from controlled combustion at high temperature (500-600°C) will produce silica ash that can be utilized in various chemical process. Controlled combustion at temperature higher than 1000°C will lead to the production of crystalline silica[9]. Silica content in rice husk ash is approximately 86% - 97% dry weight. According to chemical composition data, rice husk ash contains some essential chemical elements as shown in Table 2.

<table>
<thead>
<tr>
<th>Component</th>
<th>Weight (%)</th>
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<tbody>
<tr>
<td>SiO₂</td>
<td>86.90 – 97.30</td>
</tr>
<tr>
<td>K₂O</td>
<td>0.58 – 2.50</td>
</tr>
<tr>
<td>Na₂O</td>
<td>0.00 – 1.75</td>
</tr>
<tr>
<td>CaO</td>
<td>0.20 – 1.50</td>
</tr>
<tr>
<td>MgO</td>
<td>0.12 – 1.96</td>
</tr>
<tr>
<td>Fe₂O₃</td>
<td>0.00 – 0.54</td>
</tr>
<tr>
<td>P₂O₅</td>
<td>0.20 – 2.84</td>
</tr>
<tr>
<td>SO₃</td>
<td>0.10 – 1.13</td>
</tr>
<tr>
<td>Cl</td>
<td>0.00 – 0.42</td>
</tr>
</tbody>
</table>

Glycerol is the simplest glyceride compound with hydrophilic and hygroscopic hydroxyl. Glycerol is the component which composes various kinds of lipid includes triglyceride [11]. At low glycerol content, the constructed polymer has brittle structure which indicates weak and inflexible characteristics [12]. Theoretically, plasticizer can lower internal force among polymer chains which leads to the decrease of stiffness and improvement of moisture permeability [6].

Extrusion is continuous manufacture process used to form long products with fixed sections. This technique can be used for processing most of the thermoplastic polymer and some thermoset polymer. Generally, plastic that can be processed using extrusion method has high viscosity so that extruded product can maintain its form until fast cooling step achieved (water bath, air quench atau chill roll) [13].

3. Methodology

Materials and Equipment
The materials used in this research were rice husk and rice husk ash as fillers which are obtained from Kilang Padi Ginting JI. Tanjung selamat, Medan Tuntungan. Plastic drinking bottle waste as matrix was obtained from junk collectors around University of Sumatera Utara and Setia Budi Medan. Glycerol 99.7% as plasticizer was obtained from UD Rudang Jaya JI. Dr. Mansyur, Medan, Sumatera Utara. The equipment used was Ball Mill, Extruder, Hotpress, Dumbbell Cutter, sieve, tensile strength test equipment, Scanning Electron Microscope (SEM), and Fourier Transform Infra-Red (FTIR).

Preparation of Hybrid Composite PET Plastic Drinking Bottle Waste Filled by Rice Husk-Rice Husk Ash with The Addition of Glycerol as Plasticizer
Rice husk was milled and sieved until 100 mesh particles obtained. Rice husk was dried in the oven at 70°C [1] to reduce water content in rice husk particles. Rice husk ash was directly sieved until 100 mesh particles obtained and followed by drying process in the oven at 70°C. Plastic drinking bottle waste was washed and cut into smaller pieces using scissor.

The ratio between rice husk and rice husk ash particle was 1:1 (w/w). The ratio between plastic drinking bottle waste and rice husk- rice husk ash particles was 95:5, 90:10 and 85:15 (w/w). Glycerol addition as much as 3% of the plastic drinking bottle waste volume was 2.06 ml; 1.956 ml; dan 1.847 ml. Then they were put into beaker glass and stirred until homogeneous mixture obtained. The mixture of rice husk and rice husk ash particle was smelted in extruder at 265°C. The extruded mixture then was formed into specimen according to ASTM D 638-10. It was done by placing and pressing the filled former in hotpress at 265°C for 5 minutes. Then the cooling step was carried out at 25°C-30°C for 5-10 minutes so that the specimen was not too hot when the composite was released from its former. The composite specimens which were released form the formers were tested and characterized using tensile strength test equipment, SEM and FTIR.

4. Results and Discussion

FTIR characteristic of rice husk-rice husk ash filled plastic drinking bottle waste hybrid composite with the addition of glycerol as plasticizer can be seen in Figure 1.

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**Table 1: Chemical Composition of Rice Husk [8]**

**Table 2: Chemical Composition of Rice Husk Ash [8]**

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**Figure 1: FTIR characteristic of rice husk-rice husk ash filled plastic drinking bottle waste hybrid composite with the addition of glycerol as plasticizer**

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Figure 1: FTIR Characterization of Rice Husk-Rice Husk Ash Filled Plastic Drinking Bottle Waste Hybrid Composite with the Addition of Glycerol as Plasticizer

Figure 1 shows that the presence of wave number at 3429.43 cm⁻¹ indicates hydroxyl (-OH) functional groups from stretched alcohol of cellulose chain and lignin. Wave number at 2966.52 cm⁻¹ indicates C-H functional groups from stretched alkanes and wave number at 1720.50 cm⁻¹ indicates C=O functional groups from stretched carbonyl and acetyl of xylene component from pentose and lignin [14]. Wave number at 1099.43 cm⁻¹ indicates C-O-C functional groups from stretched ether of lignin and pentoses whereas wave number at 729.09 cm⁻¹ indicates C-H from bend aromatic ring of benzene [15].

From Figure 1, it can also be seen that there is no new absorption peak formed comparing to the FTIR characterization of PET plastic drinking bottle waste and rice husk-rice husk ash filled hybrid composite. This condition shows that the hybrid composite did not undergo any chemical reaction. The combination between the matrix and fillers was only achieved by physical reaction.

Figure 2 shows the effect of rice husk-rice husk ash with glycerol as plasticizer addition into PET plastic drinking bottle waste matrix on the tensile strength of hybrid composite.

Figure 2: The Effect of Rice Husk-Rice Husk Ash with Glycerol as Plasticizer Addition on the Tensile Strength of PET Plastic Drinking Bottle Waste Hybrid Composite

From Figure 2, it can be seen that the tensile strength of either hybrid composite or glycerol-filled hybrid composite at all ratio is lower than that of PET plastic drinking bottle waste. It is due to the addition of fillers will cause the decrease of composite's tensile strength. The tensile strength test result shows that the maximum tensile strength obtained for glycerol filled hybrid composite at 95/5 ratio is 2.717 MPa, while hybrid composite without glycerol addition has the maximum tensile strength at 90/10 ratio and get the value of 2.711 MPa. These maximum tensile strength obtained are lower than the tensile strength of PET plastic drinking bottle waste which get value of 4.658 MPa.

The decrease of tensile strength value in hybrid composite and glycerol filled hybrid composite is due to the weak bonding between hydrophobic polymer matrix and hydrophilic fillers. The agglomeration of fillers are also create an inhomogeneous distribution among matrix [16]. This condition will make the interphase area weak, thus decreasing the strength of composite material to receive stress. Besides, the addition of glycerol in hybrid composite is also contribute for lowering the tensile strength. It is due to the low molecular weight of glycerol (92.02 g/mol) which enables it to enter polymer chains easily and improves the flexibility of hybrid composite [17].

The same result has ever been reported by Shivappa et al. (2013) [18] for using different matrix and filler in reinforcing the composite at tensile strength test.

Figure 3 shows the effect of rice husk-rice husk ash filler with glycerol addition on elongation at break of PET plastic drinking bottle waste and rice husk-rice husk ash filled hybrid composite plastic drinking bottle waste.

Figure 3: The Effect of Rice Husk-Rice Husk Ash with Glycerol Addition on Elongation at Break of PET Plastic Drinking Bottle Waste Hybrid Composite

Composite testing results show that elongation at break decreases as the filler addition ratio either for hybrid composite or glycerol filled hybrid composite increases. The elongation at break decrease indicates the reduction of matrix ability to retain stress transfer from polymer material towards the fillers. It is due to the addition of rice husk-rice husk ash filler into the matrix will make a weak bond, thus reducing the elasticity of matrix which will lead to the stiffness of the composite [19]. This condition shows that the increment of fillers will lead to the stiffness of composite material.

Generally, the high elasticity of a material can be indicated from the high elongation at break value. Incorporation of fillers will cause the matrix loss its elasticity. It is due to the restriction of matrix mobility and deformability caused by filler addition [20]. However, glycerol addition into hybrid composite will lead to the improvement of its elasticity. It can be seen in Figure 3 that glycerol filled hybrid composite has higher elongation at break value than PET plastic drinking bottle waste. Glycerol addition will enhance mobility of polymer chain molecular which is shown by the improvement of composite elasticity, thus the elongation at break value tends to increase [21].

Modulus Young value of rice husk-rice husk ash filled hybrid composite with glycerol addition as plasticizer can be seen in Table 3. Modulus Young is a parameter which indicates the stiffness characteristic of a material. Low modulus young value signify the flexible material, while high modulus young value signify the stiff and rigid material [22].
Table 3 shows that the rice husk-rice husk filled hybrid composite with the addition of glycerol at 95/5 ratio has the lowest modulus young value, while the hybrid composite at 85/15 ratio has the highest modulus young value. This condition shows that the higher incorporation of rice husk-rice husk ash filler ratio into PET plastic drinking bottle waste matrix will tend to increase the stiffness and rigidity of the composite. On the other hand, glycerol addition can improve the elasticity of composite, thus lowering the stiffness and rigidity of the plastic drinking bottle waste matrix composite [21].

Fractured surface morphology characterization can be shown using Scanning Electron Microscopy (SEM) Analysis. Figure 4 shows the fractured morphology of PET plastic drinking bottle waste, rice husk-rice husk ash filled plastic drinking bottle waste hybrid composite with the addition of glycerol as plasticizer at 95/5 and 85/15 ratio with 1500x magnification.

It can be seen from Figure 4(a) that tensile strength test-fractured morphology of PET plastic drinking bottle waste has stiff and rigid structure. Figure 4(b) shows that fractured morphology of ice husk-rice husk ash filled PET plastic drinking bottle waste hybrid composite with the addition of glycerol as plasticizer at 95/5 ratio has a good filler distribution, no agglomeration and good interface bond between the matrix and fillers. Figure 4(c) shows that fractured morphology of ice husk-rice husk ash filled PET plastic drinking bottle waste hybrid composite with the addition of glycerol as plasticizer at 85/15 ratio has inhomogeneous filler distribution and the presence of void. Sedangkan Gambar 4(c) menunjukkan morfologi patahan dari Komposit Hibrid PET LBPKM- Abu Sekam Padi dan Sekam Padi dengan penambahan gliserol sebagai plasticizer pada rasio 85/15 menunjukkan bahwa penyebab pengisinya tidak merata dan terdapat fraksi kosong (void) di dalamnya. The void presence can influence the bonding between matrix and the filler particle. The presence of void in particle will cause the matrix unable to fill the empty space in the former. When the load is applied on the composite, stress area will move to void area, thus decreasing the strength of composite [23].

5. Conclusion

FTIR analysis result of PET plastic drinking bottle waste and rice husk-rice husk ash filled PET plastic drinking bottle waste hybrid composite with the addition of glycerol as plasticizer shows no significant changes in functional groups due to the absence of reaction during mixing process. The addition of glycerol as plasticizer still unable to improve the rice husk-rice husk ash filled PET plastic drinking bottle waste hybrid composite at all different ratio comparing to that of PET plastic drinking bottle waste. From the elongation at break analysis result of rice husk-rice husk ash filled PET plastic drinking bottle waste hybrid composite, elongation at break
characteristics will increase with the addition of glycerol as plasticizer, but decreasing as the filler ratio increases.

References


